

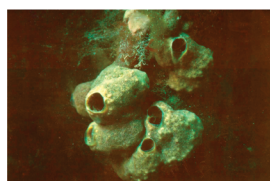
Peloruside B, A Potent Antitumor Macrolide from the New Zealand Marine Sponge *Mycale hentscheli*: Isolation, Structure, Total Synthesis, and Bioactivity

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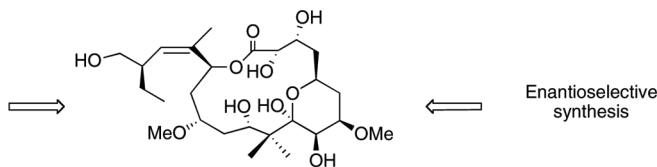
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Mycale hentscheli



Isolation, structure and synthesis of peloruside B

Peloruside B (**2**), a natural congener of peloruside A (**1**), was isolated in sub-milligram quantities from the New Zealand marine sponge *Mycale hentscheli*. Peloruside B promotes microtubule polymerization and arrests cells in the G₂/M phase of mitosis similar to paclitaxel, and its bioactivity was comparable to that of peloruside A. NMR-directed isolation, structure elucidation, structure confirmation by total synthesis, and bioactivity of peloruside B are described in this article. The synthesis features Sharpless dihydroxylation, Brown's asymmetric allylboration reaction, reductive aldol coupling, Yamaguchi macrolactonization, and selective methylation.

Introduction

The New Zealand marine sponge *Mycale hentscheli* (Poecilosclerida, Demospongiae) is the source of three classes of biologically active secondary metabolites, each with a different activity profile. Heterocyclic amides

mycalamides A, B, and D and the macrodiolide pateamine are protein synthesis inhibitors acting at different stages of translation.^{1–6} In 2000, the polyoxygenated, 16-membered macrolide peloruside A (**1**, Figure 1) was reported from a Pelorus Sound collection of *M. hentscheli*.⁷ Peloruside A has shown potent cytotoxic activity against P388 murine leukemia cells, with an IC₅₀ value of 10 ng/mL (18 nM). Furthermore, it causes cells to accumulate in the G₂/M phase of mitosis by promoting microtubule polymerization.⁸ This activity is similar to the highly successful antitumor drug paclitaxel (Taxol), and thus peloruside A has become a promising pharmaceutical candidate and a popular synthetic

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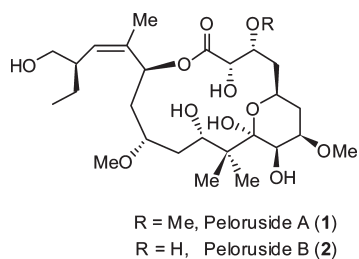


FIGURE 1. Structures of peloruside A (**1**) and peloruside B (**2**).

target. To date, a number of total syntheses^{9–13} as well as synthetic studies¹⁴ of peloruside A have been reported. Recently, a review on the synthesis and biological activity of peloruside A has appeared.¹⁵ Currently, quantities required for biological studies are still presently achieved mainly by collection from the wild and aquaculture of the sponge.^{16,17} Recently, it was shown that, like laulimalide, peloruside A occupies a different tubulin binding site than paclitaxel^{18–21} and acts synergistically with taxoid site drugs (e.g., paclitaxel, discodermolide, epothilones) in an in vitro assay using isolated tubulin as well as in cultured cells.²² Further studies have shown that peloruside A does not, in contrast to paclitaxel, induce the production of pro-inflammatory mediators in murine macrophages.²³

Our continuing interest in *M. hentscheli* and NMR-directed isolation has resulted in the isolation of a new compound, peloruside B (**2**), possessing the 3-des-*O*-methyl variant of peloruside A. Peloruside B was isolated in submilligram

quantities from a wild *M. hentscheli* specimen collected from Kapiti Island off the southwestern coast of the North Island of New Zealand. Its bioactivity was comparable to peloruside A, and it promotes microtubule polymerization and arrests cells in the G₂/M phase of mitosis as does paclitaxel. The structure of peloruside B was confirmed unequivocally through an enantioselective total synthesis and comparison of bioactivity with the natural product. Herein, we report the isolation of peloruside B, structure elucidation, structure confirmation by total synthesis, and evaluation of its bioactivity.

Results and Discussion

Isolation and Structure. Methanolic extracts of a 230 g wild Kapiti Island *M. hentscheli* specimen were fractionated using reversed- (PSDVB, Me₂CO/H₂O and MeOH/H₂O gradients) and normal-phase (DIOL, MeCN/CH₂Cl₂ and *i*-PrOH/*n*-hexane) column chromatography, yielding the known compounds mycalamides A (7.0 mg) and D (3.0 mg), peloruside A (**1**) (1.1 mg), and the new compound peloruside B (**2**) (327 μg). Positive-ion HRESIMS of **2** afforded a molecular formula of C₂₆H₄₆O₁₁ (*m/z* 557.2935, [M + Na]⁺, calcd 557.2932), which, like **1**, required four degrees of unsaturation. The ¹H NMR spectrum of **2** contained several resonances similar to the parent structure **1**, except for the observation of two methyl ether resonances (δ_H 3.38 and 3.48) rather than three, which was consistent with the difference of 14 mass units between **1** and **2**. Detailed inspection of the 1D and 2D NMR spectra (COSY, HSQC, HMBC) for **1** and **2** revealed the very similar nature of the two compounds, which allowed most ¹³C and ¹H NMR resonances to be assigned by direct comparison (Table 1), resulting in the same pelorusane carbon skeleton as represented by **1**. HMBC correlations (Figure 2) from 7-OMe (δ_H 3.38) to C-7 (δ_C 75.9) and 13-OMe (δ_H 3.48) to C-13 (δ_C 77.9) established the connection of the two oxymethyls. Changes in chemical shift (Δδ_{2–1}) for the CH-3 oxymethine (Δδ_{C-3} + 8.5 ppm; Δδ_{H-3} + 0.35 ppm) clearly showed a change in substitution at this center, confirming the absence of an oxymethyl at this position. Near identical chemical shifts, ¹H–¹H coupling constants, and diagnostic NOESY correlations (all within experimental error) between **1** and **2** strongly suggested the retention of relative configuration. Hence, peloruside B (**2**) is established here as the 3-des-*O*-methyl congener of peloruside A (**1**).

Enantioselective Synthesis of Peloruside B. Peloruside B has been isolated in miniscule quantity, and to confirm relative and absolute configuration of peloruside B, as well as to compare its bioactivity, we have undertaken an enantioselective synthesis of the proposed structure of peloruside B. Our retrosynthetic analysis of peloruside B is shown in Figure 3. Strategic bond disconnection of the macrolactone bond provided acyclic intermediate **3**. We envisioned that the presence of a *gem*-dimethyl at C-10 and the resulting lactol formed between C-5 and C-9 makes the 16-membered ring very sterically hindered for macrolactonization. Therefore, we planned to protect the hydroxyl group at C-5 and carry out a 16-membered ring macrocyclization. Further disconnection of the C-10 to C-11 bond provided segments **4** and **5**, which can be assembled by a

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TABLE 1. ^{13}C (150 MHz) and ^1H (600 MHz) NMR Data for Natural Peloruside B (2) in CDCl_3

position	^{13}C		^1H		COSY	HMBC	NOESY
	δ_{C}	mult.	δ_{H} (mult., J , Hz)				
1	174.5	C	—	—	—	—	2-OH, 4a (w), 5, 11, 15
2	69.8	CH	4.49 (s)	—	—	1	2, 5, 7, 8, 11, 21
2-OH	—	—	6.74 (br s)	—	—	—	4b, 11, 15 (w), 23
3	69.9	CH	4.57 (br s)	—	4a, 4b	—	2 (w), 4b, 5
4a	38.4	CH_2	1.82 (td, 11.7, 3.8)	—	3, 4b, 5	3	3, 4a, 6a
4b	—	—	1.98 (m)	—	3, 4a, 5	3	2, 2-OH, 4a, 6b, 7
5	63.7	CH	4.15 (tdd, 14.5, 4.4, 2.4)	—	4a, 4b, 6a	—	4b, 6b, 7
6a	31.4	CH_2	1.51 (q, 12.0)	—	5, 6b, 7	7	5, 6a, 7
6b	—	—	1.75 (ddd, 12.6, 5.0, 2.4)	—	6a, 7	7, 8	2-OH, 5, 6a, 6b, 7-OMe, 8
7	75.9	CH	3.79 (ddd, 11.4, 5.0, 3.2)	—	6a, 6b, 8	7-OMe	7, 8
7-OMe	55.7	CH_3	3.38 (s)	—	—	7, 7-OMe	2-OH, 7, 7-OMe, 21
8	66.9	CH	4.00 (d, 3.0)	—	7	6, 7, 9	—
9	101.9	C	—	—	—	—	—
10	43.6	C	—	—	—	—	2, 2-OH, 3, 12a, 14a, 22
11	73.8	CH	4.91 (br d, 12.0)	—	12b	—	11, 12b, 13, 22
12a	33.9	CH_2	1.42 (m)	—	12b, 13	—	12a, 21
12b	—	—	2.07 (m)	—	11, 12a, 13	—	12a, 12b, 13-OMe, 14b
13	77.9	CH	3.98 (m)	—	12a, 12b, 14a, 14b	—	13, 15
13-OMe	59.1	CH_3	3.48 (s)	—	—	13, 13-OMe	11, 14b, 15, 23
14a	35.6	CH_2	2.05 (dd, 15.0, 10.8)	—	13, 14b, 15	12, 13, 15, 16	14a, 15, 20
14b	—	—	2.15 (dd, 15.0, 10.2)	—	13, 14a, 15	12, 13	2, 3 (w), 13-OMe, 14a, 14b, 17, 18, 20, 23
15	71.7	CH	5.69 (d, 11.2)	—	14a, 14b, 17, 23	1, 13, 14, 16, 17, 23	—
16	136.7	C	—	—	—	—	15, 18, 19a, 19b, 20, 23, 24a
17	130.9	CH	5.02 (d, 10.6)	—	15, 18, 23	15, 18, 19, 23, 24	15, 17, 19b, 20, 24b
18	43.3	CH	2.56 (m)	—	17, 19a, 19b, 24a, 24b	16, 17, 19, 20, 24	17, 19b, 20
19a	24.6	CH_2	1.14 (m)	—	18, 19b, 20	17, 18, 20, 24	17 (w), 18, 19a, 20, 24b
19b	—	—	1.42 (m)	—	18, 19a, 20	17, 18, 20, 24	14b, 15, 17, 18, 19a, 19b, 24b
20	12.3	CH_3	0.85 (t, 7.5)	—	19a, 19b	18, 19, 20	8, 12b
21	15.7	CH_3	1.08 (s)	—	—	9, 10, 11, 21, 22	11, 12a
22	20.8	CH_3	1.10 (s)	—	—	9, 10, 11, 21, 22	3, 15, 17
23	17.6	CH_3	1.71 (d, 1.2)	—	15, 17	15, 16, 17, 23	17, 19a, 19b, 24b
24a	67.0	CH_2	3.34 (t, 10.3)	—	18, 24b	17, 18, 19	18, 19b, 20, 24a
24b	—	—	3.62 (br d, 10.3)	—	18, 24a	17, 18, 19	—

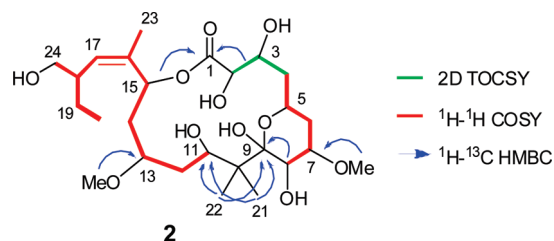


FIGURE 2. Key COSY, 2D TOCSY, and HMBC correlations for peloruside B (2).

stereoselective aldol reaction to provide **3**. Due to the steric hindrance of the *gem*-dimethyl group adjacent to the C-9 ketone, selective enolization may be problematic. We planned the same reductive aldol coupling protocol that was utilized during the synthesis of (+)-peloruside A.¹¹ Segment **4** has been synthesized by us previously,¹¹ which features asymmetric alkylation of oxazolidinone **6**, Horner–Emmons olefination, and Brown’s asymmetric allylboration. The synthesis of segment **5** can be derived from diethyl tartrate **7**.

The synthesis of C-1 to C-10 fragment **5** commenced with commercially available diethyl *D*-tartrate as shown in Scheme 1. Both hydroxyl groups were protected with MOM ether, and the resulting diester was reduced by LAH reduction to give the corresponding diol.²⁴ Treatment of the

diol with NaH and PMBCl gave monoprotected PMB ether **8** (83% yield over three steps). It was converted to alkene **9** by transformation of the alcohol into an iodide, using iodine and triphenylphosphine in THF at 0 °C, followed by substitution of the corresponding iodide with vinyl cuprate (63% yield for two steps).²⁵ The terminal olefin was converted to an aldehyde using oxidative cleavage. This was accomplished by dihydroxylation with OsO₄ followed by treatment of the resulting diol with NaIO₄.²⁶ The aldehyde was subjected to Brown’s asymmetric allylboration²⁷ to afford the corresponding homoallylic alcohol. The resulting alcohol was protected as TBS ether **10**. The allylboration proceeded with high diastereoselectivity (95:5 dr), and the diastereomers were separated at this step (53% over four steps, 95:5 dr). Conversion of terminal alkene to the corresponding aldehyde, using the above oxidative cleavage followed by Ando’s modified Horner–Emmons reaction,²⁸ provided the *Z*-olefin **11** (*Z*:*E* 7:1, 81% yield over three steps). Sharpless asymmetric dihydroxylation of the pure *Z*-olefin **11** furnished the corresponding diol in 77% yield (9:1 dr).²⁹ The resulting diol was protected as an isopropylidene derivative **12** in near quantitative yield. Reduction of

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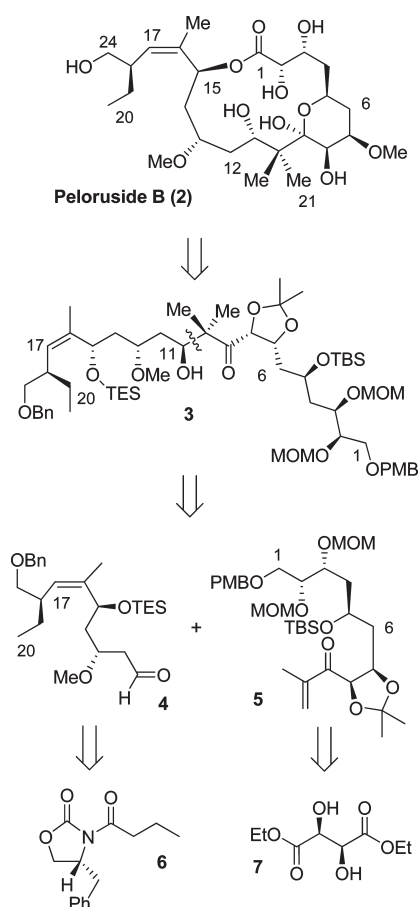


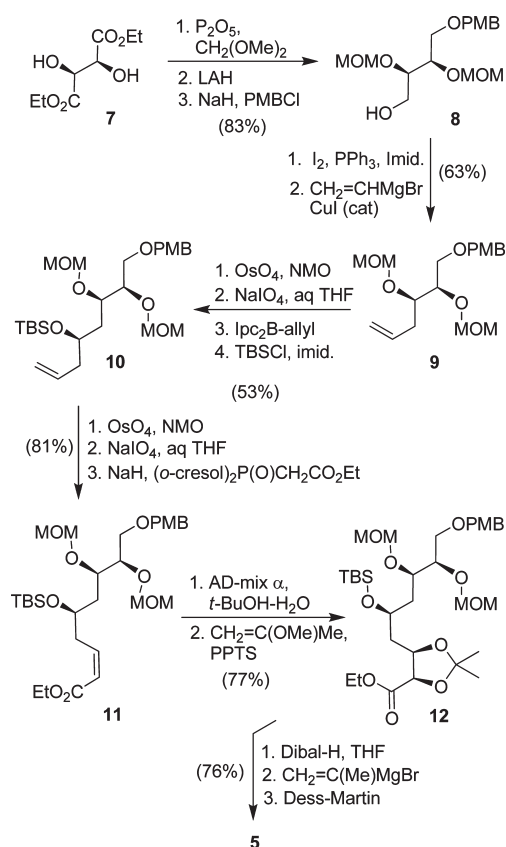
FIGURE 3. Retrosynthetic analysis of peloruside B (2).

ester **12** to the aldehyde, followed by reaction with isopropenylmagnesium bromide and Dess–Martin oxidation, provided the enone **5** in 76% yield over three steps. The other coupling fragment **4** was synthesized following the reaction schemes reported by us in the context of the synthesis of (+)-peloruside A.¹¹

With segments **4** and **5** in hand, our subsequent plan was to use a reductive aldol reaction to couple these segments and set the C-11 hydroxyl stereocenter. As shown in Scheme 2, enone **5** was treated with L-selectride to generate the corresponding enolate, which was reacted with aldehyde **4** to afford the aldol product **3** as the major desired isomer (66% yield, 6.5:1 dr by ¹H NMR). This reductive coupling reaction provided a lower yield and slightly improved diastereoselectivity compared to the C3-OMe derivative during the synthesis of peloruside A.¹¹ The stereochemical outcome and diastereomeric ratios were consistent with our previous observation.^{11,30} The TES protecting group was selectively removed by treatment with dilute TBAF at $-30\text{ }^{\circ}\text{C}$. DDQ oxidation effected the removal of the PMB group to provide alcohol **13** in 85% yield. Conversion of the primary alcohol to the *seco*-acid **14** was carried out in a two-step sequence involving (1) TPAP oxidation of the alcohol to the corresponding aldehyde and (2) oxidation of the aldehyde to the acid with NaClO₂. However, the yield was extremely low

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SCHEME 1. Synthesis of Fragment 5

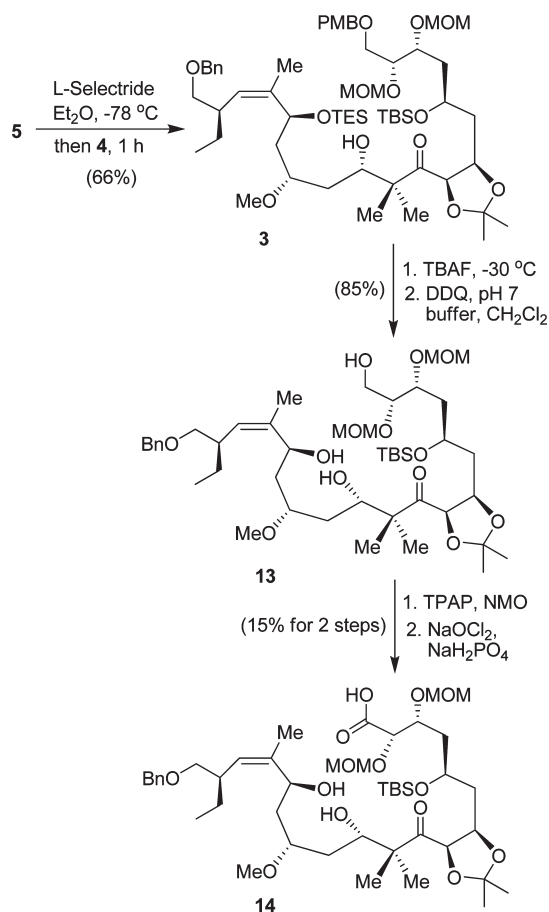


(15%). The reason for the low yield is possibly due to the presence of the C-15 allylic alcohol, which gets readily oxidized. Attempts with other oxidizing reagents were unsuccessful.

We then envisaged that selective oxidation of the primary alcohol in the presence of two other hydroxyl groups would be very difficult due to the C-15 allylic alcohol. While we achieved the selective oxidation during the synthesis of (+)-peloruside A, for peloruside B synthesis, it was difficult to selectively oxidize the primary alcohol in good yield. We decided to protect both C-11 and C-15 secondary hydroxyl groups as TES-ethers before oxidizing the primary alcohol. Reductive aldol product **3** was treated with TESOTf and pyridine, as shown in Scheme 3. Subsequent removal of the PMB protecting group with DDQ gave the corresponding primary alcohol **15** in 65% yield for two steps. The primary alcohol **15** was oxidized to the corresponding aldehyde and, without purification, further oxidized to the corresponding acid in 88% yield over two steps. Both TES protecting groups at the C-11 and C-15 positions were selectively removed with dilute HF/pyridine in 80% yield to afford the corresponding *seco*-acid **14**. The *seco*-acid **14** was subjected to the Yamaguchi protocol³¹ using 2,4,6-trichlorobenzoyl chloride in the presence of DMAP to generate macrolactone **16** in 77% yield.

The synthesis of the proposed structure of peloruside B is shown in Scheme 4. We initially attempted the

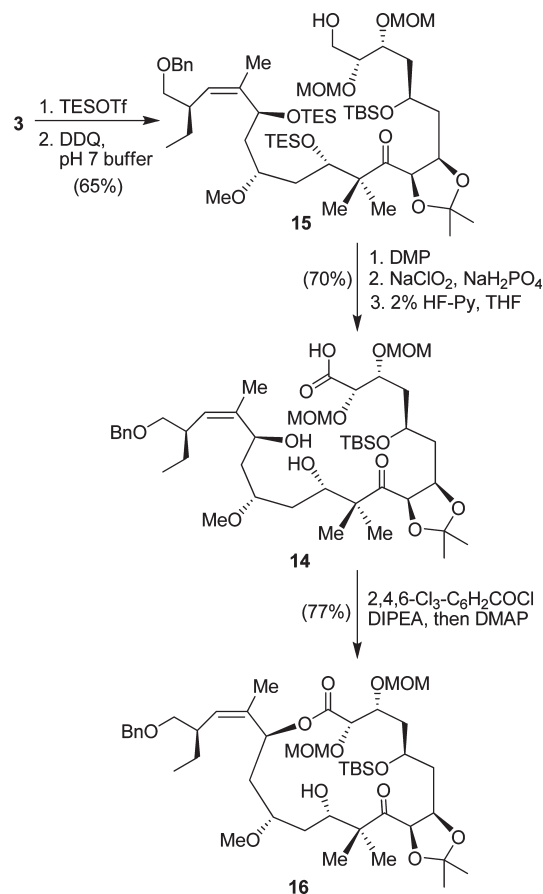
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SCHEME 2. Reductive Aldol Reaction and Synthesis of *seco*-Acid 14

deprotection of both TBS ethers and acetonide groups by treatment of **16** with 1 N HCl. However, the reaction was not very satisfactory, yielding a number of undesired products. We then planned to remove the TBS ethers of macrolactone **16** by treatment with HF/pyridine. This resulted in a mixture (6:1) of hemiketal **17a** and the corresponding hydroxyketone **17b** in 86% combined yield. The isopropylidene group from both hemiketal **17a** and hydroxyketone **17b** was deprotected with 80% aqueous acetic acid at 50 °C for 3 h to afford the corresponding diol **18** in 52% isolated yield. Selective methylation of the diol was carried out with Meerwein's reagent at -5 °C to afford the corresponding C-7 monomethylated product. The product was then exposed to 4 N HCl in THF, and this effected the removal of both MOM groups. Catalytic transfer hydrogenation with Pd/C in the presence of formic acid selectively removed the benzyl group in the presence of the C-17-trisubstituted olefin to generate the synthetic peloruside B in 24% yield over three steps. The 1D and 2D NMR data of synthetic peloruside B are in complete agreement with that of natural peloruside B.

Furthermore, using synthetic and natural peloruside B, we have performed MTT cell proliferation assays (96 h incubation) and flow cytometric measurements (16 h incubation) as previously described with human myeloid leukemic cells (HL-60 cells).²³ Natural peloruside B (**2**)

SCHEME 3. Revised Route to Macrolactone 16



gave an IC_{50} of 33 ± 10 nM, a value approximately one-third of that obtained with peloruside A (**1**) tested at the same time (10 ± 4 nM) ($n = 3$ independent experiments, Figure 4). Natural peloruside B was also slightly less active than peloruside A against arresting cells in the G_2/M phase of the cell cycle, with $56 \pm 1\%$ of cells in G_2/M for **2** ($n = 6$ replicates from three experiments) compared to $67 \pm 3\%$ in G_2/M for **1** (200 nM concentration of **1** or **2**; $n = 5$ replicates in three experiments). Control, untreated HL-60 cells had $23 \pm 2\%$ of the cells in the G_2/M phase of the cell cycle.

Synthetic **2** and natural **2** were compared in paired MTT cell proliferation assays in human ovarian carcinoma 1A9 cells. 1A9 cells were treated for 72 h with drug. The synthetic **2** (48 ± 11 nM IC_{50}) was slightly more potent than the natural **2** (71 ± 6 nM) in this cell line ($n = 3$ experiments). The G_2/M arrest results for the synthetic product were also slightly higher than those obtained with the natural congener. Synthetic **2** gave 28, 27, 43, and 70% of cells in G_2/M at concentrations of 0, 40, 100, and 140 nM, respectively (Figure 5). Natural **2** gave 25, 28, 31, and 47% at concentrations of 0, 32, 79, and 119 nM, respectively. The slight differences in biological activity between natural and synthetic **2** are well within the normal uncertainties of the assays performed. Considering the fact that the synthetic enantiomer of peloruside A has no bioactivity, the observed potent activity of synthetic peloruside B further confirms the absolute configuration of natural peloruside B.⁹

SCHEME 4. Completion of Total Synthesis of Peloruside B (2)

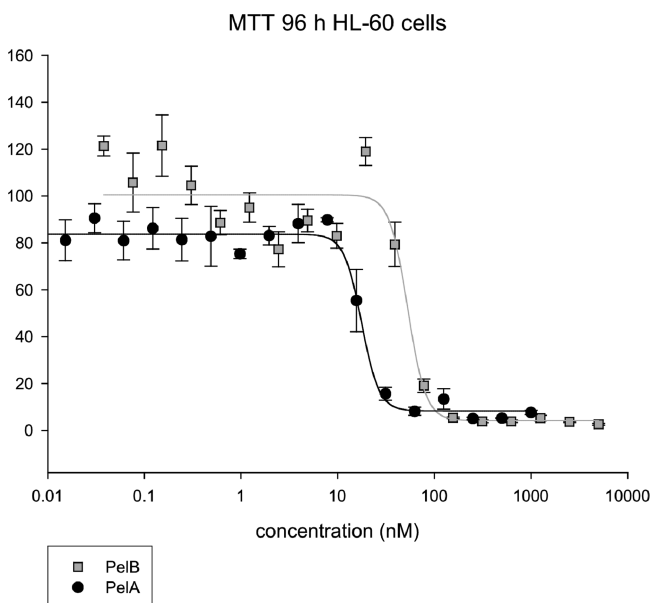
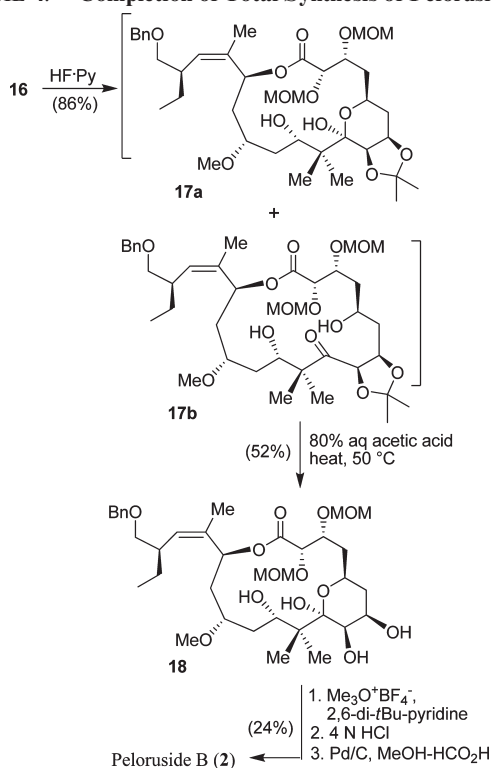


FIGURE 4. Typical MTT cell proliferation assay for pelorusides A (1) and B (2). HL-60 cells were treated for 96 h ($n = 3$ replicates from a single preparation).

Conclusion

We have isolated a sub-milligram amount of a new natural product, peloruside B (2), a natural congener of peloruside A (1), from the New Zealand marine sponge *Mycale hentscheli*. The structure of peloruside B was elucidated based upon extensive 1D and 2D NMR studies. Like peloruside A, peloruside B promotes microtubule polymerization

and arrests cells in the G₂/M phase of the cell cycle, as does paclitaxel. The bioactivity of natural peloruside B was comparable to that of peloruside A. We have accomplished an enantioselective and convergent synthesis of peloruside B. The synthesis involved Sharpless dihydroxylation, Brown's asymmetric allylboration reaction, reductive aldol coupling, and Yamaguchi macrolactonization as the key reactions. The relative and absolute configuration of 2 was confirmed by total synthesis and comparison of its bioactivity with natural peloruside B. Synthesis and biological evaluation of structural variants of peloruside B are the subjects of ongoing research in our laboratories.

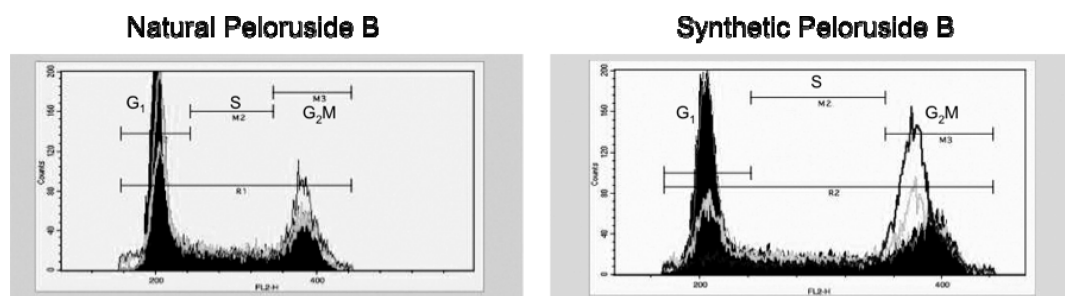
Experimental Section

Natural Peloruside B. General: The 600 MHz NMR spectra for both natural and synthetic peloruside B (2) were obtained using the same spectrometer equipped with a triple resonance HCN cryogenic probe, operating at 600 or 150 MHz for ¹H and ¹³C nuclei, respectively. Chemical shifts δ (ppm) were referenced to the residual solvent peak (δ_{H} 7.26 ppm, δ_{C} 77.16 ppm for CDCl₃). Catalytic amounts (1–2 μ L) of pyridine-*d*₅ were added to each NMR sample performed in CDCl₃ to prevent compound degradation due to trace acidity associated with the solvent. High-resolution positive-ion mass spectra were recorded on a TOF electrospray mass spectrometer. Normal-phase column chromatography was carried out using 2,3-dihydroxypropoxypropyl-derivatized silica (DIOL). Reversed-phase column chromatography was achieved using HP20 or Amberchrom poly(styrene divinylbenzene) (PSDVB) chromatographic resin. HPLC was performed using a solvent delivery module equipped with 25 mL pump heads. Solvents used for flash normal- and reversed-phase column chromatography were of HPLC or analytical grade quality. All other solvents were purified by glass distillation. Solvent mixtures are reported as % v/v unless otherwise stated. Specimens were stored at –20 °C until required.

Isolation of Natural Peloruside B: *Mycale hentscheli* (230 g, NIWA no. MNP 0026), collected at a depth of 23 m from Kapiti Island, New Zealand, was cut into small segments and extracted with MeOH (2 \times 700 mL) for 24 h. The combined extracts were loaded on to HP20 PSDVB beads, washed with H₂O, and eluted with (i) 20% Me₂CO/H₂O, (ii) 55% Me₂CO/H₂O, (iii) 55% Me₂CO/0.2 M NH₄OH, and (iv) 55% Me₂CO/0.2 M NH₄OH adjusted to pH 4 with AcOH. Fraction (ii) was concentrated to dryness to yield 82.0 mg of a viscous brown oil. The resulting oil was dissolved in MeOH, loaded onto Amberchrom PSDVB, and eluted with increasing concentrations of MeOH in H₂O (10–100%). The 48–54% MeOH/H₂O fractions were concentrated to dryness to yield mycalamide D (3 mg). The 56–60% MeOH/H₂O fractions were concentrated to dryness to yield peloruside A (1) (1.1 mg), and concentration of the 68–72% MeOH/H₂O fractions gave mycalamide A (7.0 mg). The 54–56% MeOH/H₂O fractions were concentrated to dryness to yield 5.0 mg of brown oil. A portion of this oil (3.0 mg) was recycled twice on DIOL with increasing concentrations of MeCN in CH₂Cl₂ (1–10%) and 50% MeOH/CH₂Cl₂. The 10% MeCN/CH₂Cl₂ and 50% MeOH/CH₂Cl₂ fractions were concentrated to dryness to yield a pale yellow oil (1.5 mg). This oil was then purified using HPLC (DIOL, 5 μ m, 4 mm \times 250 mm), with 20% *i*-PrOH/*n*-hexane as the mobile phase, collecting fractions at 1 min intervals to give 13.9 μ g of mycalamide D, 9.4 μ g of peloruside A (1), and 327 μ g of peloruside B (2).

Peloruside A (1): Colorless film; all other data as previously published.⁷

Peloruside B (2): Colorless film; $[\alpha]_{\text{D}}^{25}$ could not be determined accurately because of very small quantity of natural



As peloruside B was increased in concentration, more and more cells accumulated in G₂M of the cell cycle, indicating a block at mitosis. Cells were treated for 16 hr with drug, and their DNA was then stained with propidium iodide and the cells analyzed in a flow cytometer. G₁ = M1, S = M2, and G₂M = M3. Synthetic **2** gave 28%, 27%, 43% and 70% of cells in G₂M at concentrations of 0, 40, 100, and 140 nM, respectively. Natural **2** gave 25%, 28%, 31%, and 47% at concentrations of 0, 32, 79, and 119 nM, respectively.

FIGURE 5. G₂/M arrest of 1A9 cells by natural and synthetic peloruside B (**2**).

sample and magnitude of rotation was very small; NMR data see Table 1; HRESIMS, $[M + Na]^+$, observed m/z 557.29356, calculated 557.29323 for C₂₆H₄₆O₁₁Na, $\Delta = 0.58$ ppm.

Synthetic Peloruside B. General experimental details are provided as Supporting Information.

Aldol Product 3: To the enone **5** (521 mg, 0.81 mmol) dissolved in anhydrous ethyl ether (132 mL) was added L-selectride (1 M in THF, 0.85 mL, 0.85 mmol) slowly at -78 °C. The solution was stirred at -78 °C for 0.5 h to form the corresponding enolate. The formation of the enolate was monitored by TLC. Then the aldehyde **4** (500 mg, 1.12 mmol) dissolved in Et₂O (10 mL) was added dropwise over 15 min at -78 °C. The solution was stirred at -78 °C for 1 h and quenched with saturated ammonium chloride solution. The aqueous layer was washed and extracted with ethyl ether. The organic extracts were washed with brine and dried over anhydrous sodium sulfate. The solvent was removed under vacuum, and the residue was purified by flash chromatography on silica gel (hexane/ethyl acetate, 85/15) to afford the alcohol **3** (503.3 mg, 57% yield) and the other C-11 diastereomer (77 mg, 9% yield; total yield 66%, 6.5:1 dr). Alcohol **3**: ¹H NMR (400 MHz, CDCl₃) δ 0.06 (s, 3H), 0.07 (s, 3H), 0.55 (q, $J = 7.9$ Hz, 6H), 0.83 (t, $J = 7.5$ Hz, 3H), 0.87 (s, 9H), 0.89 (t, $J = 8.9$ Hz, 9H), 1.10 (s, 3H), 1.17 (s, 3H), 1.17–1.26 (m, 3H), 1.33 (s, 3H), 1.42–1.49 (m, 2H), 1.52 (s, 3H), 1.59–1.74 (m, 6H), 1.70 (s, 3H), 2.06 (m, 1H), 2.54 (m, 1H), 3.26–3.39 (m, 3H), 3.29 (s, 3H), 3.35 (s, 3H), 3.36 (s, 3H), 3.52 (dd, $J = 10.0, 7.1$ Hz, 1H), 3.57 (m, 1H), 3.63 (dd, $J = 10.0, 3.6$ Hz, 1H), 3.71 (m, 1H), 3.79 (s, 3H), 3.79 (m, 1H), 3.98 (m, 1H), 4.06 (d, $J = 10.5$ Hz, 1H), 4.45 (s, 2H), 4.48 (s, 2H), 4.55 (m, 1H), 4.62 (s, 2H), 4.65 (d, $J = 6.8$ Hz, 1H), 4.77 (d, $J = 6.8$ Hz, 1H), 4.93 (d, $J = 10.2$ Hz, 1H), 5.09 (d, $J = 6.8$ Hz, 1H), 6.86 (d, $J = 8.6$ Hz, 2H), 7.24 (d, $J = 8.6$ Hz, 2H), 7.26 (m, 5H); ¹³C NMR (100 MHz, CDCl₃) δ -4.8, -4.3, 4.7, 6.9, 11.6, 17.9, 18.0, 20.9, 25.3, 25.6, 25.8, 27.4, 33.5, 38.9, 39.1, 39.3, 51.5, 55.1, 55.7, 55.9, 56.2, 66.4, 67.3, 69.8, 72.7, 72.9, 73.0, 73.2, 74.2, 74.8, 76.4, 77.6, 79.2, 96.8, 109.2, 113.6, 127.2, 127.37, 127.45, 128.2, 129.2, 130.3, 138.5, 138.9, 159.0, 210.5; FT-IR (film, NaCl) 1039.5, 1082.5, 1249.4, 1379.1, 1463.4, 1514.0, 1613.3, 1715.8 cm⁻¹; $[\alpha]_D^{20} -7.7$ (c 1.37, CHCl₃); MS, m/z (MALDI), 1113 $[M + Na]^+$.

Primary Alcohol 15: To the alcohol **3** (435 mg, 0.4 mmol) dissolved in CH₂Cl₂ (8 mL) were added anhydrous pyridine (0.32 mL, 4 mmol) and triethylsilyl trifluoromethanesulfonate (0.18 mL, 0.8 mmol) sequentially at 0 °C. The mixture was stirred at 0 °C for 2 h and quenched with a saturated NaHCO₃ solution. The aqueous layer was extracted with CH₂Cl₂ (3 \times 10 mL). The combined organic extracts were washed with brine and dried over anhydrous Na₂SO₄. The solvent was evaporated under vacuum, and the residue was purified by flash

chromatography on silica gel (hexanes/ethyl acetate 90/10) to afford the TES protected product (425 mg, 88% yield): ¹H NMR (400 MHz, CDCl₃) δ 0.06 (s, 3H), 0.08 (s, 3H), 0.58 (q, $J = 7.9$ Hz, 6H), 0.68 (q, $J = 7.9$ Hz, 6H), 0.82 (t, $J = 7.4$ Hz, 3H), 0.88 (s, 9H), 0.91 (t, $J = 7.9$ Hz, 9H), 1.00 (t, $J = 7.9$ Hz, 9H), 1.09 (s, 3H), 1.13 (s, 3H), 1.23–1.38 (m, 2H), 1.49 (s, 3H), 1.51–1.73 (m, 4H), 1.70 (s, 3H), 1.98 (m, 1H), 2.58 (m, 1H), 3.21 (s, 3H), 3.27–3.40 (m, 2H), 3.34 (s, 3H), 3.37 (s, 3H), 3.47 (m, 1H), 3.53 (dd, $J = 10.0, 6.9$ Hz, 1H), 3.64 (dd, $J = 10.0, 3.7$ Hz, 1H), 3.68 (m, 1H), 3.80 (s, 3H), 3.81 (m, 1H), 3.97 (m, 1H), 4.07 (d, $J = 9.2$ Hz, 1H), 4.46 (s, 2H), 4.50 (s, 2H), 4.51 (m, 1H), 4.57 (m, 1H), 4.62 (s, 2H), 4.65 (d, $J = 6.8$ Hz, 1H), 4.78 (d, $J = 6.8$ Hz, 1H), 4.97 (d, $J = 10.2$ Hz, 1H), 5.09 (d, $J = 6.7$ Hz, 1H), 6.86 (d, $J = 8.6$ Hz, 2H), 7.24 (d, $J = 8.6$ Hz, 2H), 7.32 (m, 5H); ¹³C NMR (100 MHz, CDCl₃) δ -4.9, -4.3, 4.8, 5.5, 6.9, 7.1, 11.4, 17.89, 17.93, 19.7, 21.7, 25.2, 25.5, 25.8, 27.7, 38.4, 38.9, 39.4, 39.8, 52.7, 55.1, 55.7, 55.8, 66.3, 67.1, 69.7, 72.90, 72.93, 73.0, 73.2, 74.1, 74.4, 74.9, 77.6, 80.7, 96.9, 108.7, 113.6, 127.3, 127.5, 127.6, 128.2, 129.2, 130.3, 138.5, 138.6, 159.0, 209.2; FT-IR (film, NaCl) 1039.8, 1100.9, 1248.5, 1379.2, 1462.3, 1513.8, 1612.4, 1718.3 cm⁻¹; $[\alpha]_D^{20} -10.1$ (c 1.1, CHCl₃); MS, m/z (ESI), 1228 $[M + Na]^+$.

The TES protected product from above (360 mg, 0.3 mmol) was dissolved in CH₂Cl₂ (60 mL), and pH 7 buffer (12 mL) was added. DDQ (136 mg, 0.6 mmol) was added while the mixture was vigorously stirred. The mixture was stirred for 4 h before another portion of DDQ (136 mg, 0.6 mmol) and pH 7 buffer (12 mL) was added. The mixture was further stirred for 4 h. Then, a saturated aqueous NaHCO₃ solution was added, the aqueous layer was extracted with CH₂Cl₂ (3 \times 50 mL), and the combined extracts were washed with saturated NaHCO₃ solution and brine and dried over anhydrous Na₂SO₄. The solvent was evaporated under vacuum, and the residue was purified by column chromatography on silica gel (hexanes/ethyl acetate 2/1) to afford the primary alcohol **15** (240 mg, 74% yield) as a colorless oil: ¹H NMR (400 MHz, CDCl₃) δ 0.10 (s, 3H), 0.11 (s, 3H), 0.58 (q, $J = 7.9$ Hz, 6H), 0.70 (q, $J = 7.9$ Hz, 6H), 0.84 (t, $J = 7.4$ Hz, 3H), 0.91 (s, 9H), 1.01 (t, $J = 7.9$ Hz, 9H), 1.11 (s, 3H), 1.15 (s, 3H), 1.27–1.41 (m, 3H), 1.32 (s, 3H), 1.51 (s, 3H), 1.53–1.65 (m, 2H), 1.69 (s, 3H), 1.75 (m, 2H), 2.02 (m, 1H), 2.59 (m, 1H), 3.17 (m, 1H), 3.24 (s, 3H), 3.31 (dd, $J = 16.0, 5.6$ Hz, 1H), 3.38 (dd, $J = 9.3, 5.6$ Hz, 1H), 3.41 (s, 3H), 3.43 (s, 3H), 3.49 (m, 1H), 3.65 (m, 2H), 3.75 (m, 2H), 3.97 (m, 1H), 4.09 (d, $J = 9.3$ Hz, 1H), 4.49 (m, 1H), 4.50 (s, 2H), 4.60 (dd, $J = 8.0, 4.7$ Hz, 1H), 4.64 (d, $J = 6.8$ Hz, 1H), 4.68 (d, $J = 7.9$ Hz, 2H), 4.73 (d, $J = 6.8$ Hz, 1H), 4.98 (d, $J = 10.2$ Hz, 1H), 5.12 (d, $J = 6.7$ Hz, 1H), 7.28 (m, 5H); ¹³C NMR (125 MHz, CDCl₃) δ -4.9, -4.3, 4.8, 5.5, 6.9, 7.0, 11.4, 17.88, 17.93, 19.7, 21.6, 25.2, 25.5, 25.7, 27.7, 38.4, 39.0, 39.1, 39.5, 39.8, 52.7, 52.8,

55.8, 55.9, 62.1, 66.2, 67.1, 73.0, 73.2, 74.1, 74.4, 75.4, 80.5, 82.4, 97.1, 97.6, 108.7, 127.3, 127.5, 127.6, 128.2, 138.5, 138.6, 209.2; FT-IR (film, NaCl) 1037.5, 1082.4, 1249.1, 1379.4, 1462.8, 1716.7 cm^{-1} ; $[\alpha]_{\text{D}}^{20}$ -3.5 (c 1.02, CHCl_3); MS, m/z (MALDI), 1108 $[\text{M} + \text{Na}]^+$.

Macrolactone 16: To the primary alcohol **15** from above (245 mg, 0.226 mmol) dissolved in CH_2Cl_2 (12 mL) were added NaHCO_3 (38 mg, 0.45 mmol) and Dess–Martin periodinane (144 mg, 0.34 mmol). The mixture was stirred at 23 °C for 1 h and quenched with a saturated aqueous NaHCO_3 solution (10 mL) and a saturated aqueous sodium thiosulfate solution (10 mL). The mixture was stirred until the two layers became clear. The aqueous layer was extracted with CH_2Cl_2 (3×10 mL), and the combined extracts were washed with brine and dried over anhydrous Na_2SO_4 . The solvent was evaporated under vacuum, and the crude aldehyde was used for the next step without purification.

The crude aldehyde from above was dissolved in *t*-BuOH (8 mL) and H_2O (2 mL). To this solution were added 2-methyl-2-butene (2 mL) followed by sodium chlorite (213 mg) and sodium phosphate monobasic monohydrate (213 mg) in H_2O (2.2 mL) dropwise. The mixture was stirred at 23 °C for 25 min. Then, ethyl acetate was added, and the organic layer was washed with brine and dried over anhydrous Na_2SO_4 . The solvent was removed under vacuum. The crude acid was passed through a short column to afford the crude acid (219 mg, 88% yield, over two steps).

To the acid from above dissolved in THF (15 mL) was added 10% HF·Py in THF solution (1 mL of HF·Py complex dissolved in 9 mL of THF, 3 mL) at 0 °C. The solution was allowed to warm to 23 °C for 2 h. Then, more 10% HF·Py (3 mL) was added at 23 °C. The mixture was kept stirring at 23 °C for another 9 h. TLC evidence showed that the two TES groups were removed. The mixture was cooled to 0 °C, and 1 mL of H_2O was added. Lithium carbonate was used to quench the reaction, and the mixture was dried over anhydrous Na_2SO_4 . The solid was filtered and washed with ethyl ether. The solvent was evaporated under vacuum, and the crude product was passed through a short silica gel column to afford the crude acid **14** (138 mg, 80% yield) as a white solid.

To the *seco*-acid **14** (115 mg, 0.132 mmol) from above dissolved in THF (4 mL) were added diisopropylethyl amine (172 μL , 1 mmol) and 2,4,6-trichlorobenzoyl chloride (103 μL , 0.66 mmol). The mixture was stirred at 23 °C for 4 h. Then most solvent was removed under vacuum followed by addition of dry toluene (25 mL). The mixed anhydride toluene solution was added to the DMAP (403 mg, 3.3 mmol) solution in toluene (200 mL) via a syringe pump over 13 h at 23 °C. White precipitate was formed during the addition. The mixture was stirred for an additional 36 h at 23 °C. Then, 0.1 N HCl was added to quench the reaction at 0 °C, the aqueous layer was extracted with ethyl acetate, and the combined extracts were washed with saturated NaHCO_3 solution and brine and dried over Na_2SO_4 . The solvent was evaporated under vacuum, and the crude product was purified by column chromatography on silica gel (hexanes/ethyl acetate 85/15) to afford macrolactone **16** (87 mg, 77% yield) as a colorless oil: ^1H NMR (500 MHz, CDCl_3) δ 0.09 (s, 3H), 0.10 (s, 3H), 0.86 (t, $J = 7.5$ Hz, 3H), 0.89 (s, 9H), 1.20 (s, 3H), 1.23–1.33 (m, 2H), 1.36 (s, 3H), 1.43 (s, 3H), 1.59 (s, 3H), 1.59–1.71 (m, 4H), 1.70 (s, 3H), 1.91 (m, 1H), 2.04 (m, 2H), 2.81 (m, 1H), 3.24 (d, $J = 11.4$ Hz, 1H), 3.31 (s, 3H), 3.34 (s, 3H), 3.36 (s, 3H), 3.55 (dd, $J = 9.2, 4.2$ Hz, 1H), 3.73–3.79 (m, 2H), 3.99 (d, $J = 6.4$ Hz, 1H), 4.09 (s, 1H), 4.51 (dd, $J = 25.2, 12.2$ Hz, 2H), 4.61–4.68 (m, 4H), 4.79 (d, $J = 6.9$ Hz, 1H), 4.92 (d, $J = 7.3$ Hz, 1H), 5.10 (d, $J = 10.0$ Hz, 1H), 5.89 (dd, $J = 6.8, 3.1$ Hz, 1H), 7.33 (m, 5H); ^{13}C NMR (125 MHz, CDCl_3) δ $-5.0, -4.2, 11.7, 18.0, 18.2, 21.1, 24.9, 25.2, 25.4, 25.9, 26.7, 29.7, 35.9, 38.1, 38.2, 39.3, 41.4, 50.4, 56.2, 57.5, 65.7, 70.4, 72.8, 72.9, 73.9, 75.6, 80.2, 96.2, 97.3, 110.2, 127.2,$

127.5, 128.2, 130.7, 133.8, 139.0, 169.2, 215.5; FT-IR (film, NaCl) 1024.8, 1102.2, 1257.9, 1378.4, 1456.7, 1733.0 cm^{-1} ; $[\alpha]_{\text{D}}^{20}$ -12 (c , 0.8, CHCl_3); HRMS (ESI) $[\text{M} + \text{Na}]^+$ calcd for $\text{C}_{45}\text{H}_{76}\text{O}_{13}\text{SiNa}$ 875.4953, found 875.4951.

Peloruside B (2): To the macrolactone **16** (64.7 mg, 0.08 mmol) dissolved in THF (0.5 mL) in a small plastic vial was added HF·Py complex (HF·Py/Py/THF = 1:2:4, 2 mL) at 0 °C. The mixture was allowed to warm to 23 °C and stirred overnight. Then, THF (10 mL) was added and the solution was cooled to 0 °C. The diluted solution was added to a chilled saturated NaHCO_3 solution dropwise. The aqueous layer was extracted with ethyl acetate (3×15 mL). The combined extracts were washed with brine and dried over anhydrous Na_2SO_4 . The solvent was evaporated under vacuum, and the crude product was purified by flash column chromatography on silica gel (hexanes/ethyl acetate 8/2 to 3/2 to 1/1 to 2/8) to afford hemiketal **17a** (41.2 mg) and a hydroxyl ketone **17b** (7 mg, total 48.2 mg, 86% yield).

The hemiketal **17a** (31 mg, 0.042 mmol) was dissolved in acetic acid (8/2 v/v, 2 mL) and heated to 50 °C for 3 h. After cooling to 23 °C, the solvent was removed under vacuum. The crude product was purified by column chromatography on silica gel ($\text{CHCl}_3/\text{MeOH}$ 95/5) to afford the diol (15.7 mg, 52% yield) along with the recovered starting material (10 mg).

To the diol from above (15.7 mg, 0.02 mmol) dissolved in CH_2Cl_2 (6 mL) were added molecular sieves (400 mg), 2,6-di-*tert*-butylpyridine (100 μL , 0.45 mmol), and $\text{Me}_3\text{O}^+\text{BF}_4^-$ (33.3 mg, 0.22 mmol) at -5 °C. The mixture was stirred at -5 °C for 18 h. The molecular sieves were filtered and washed with CH_2Cl_2 . The filtrate was washed with saturated NaHCO_3 solution and brine and dried over anhydrous Na_2SO_4 . The solvent was evaporated under vacuum, and the crude product was purified by flash column chromatography on silica gel ($\text{CHCl}_3/\text{MeOH}$ 98/2) to afford the crude monomethylated product (6.1 mg) along with the recovered starting material (8.7 mg).

The crude monomethylated product was dissolved in THF (1.5 mL) and cooled to 0 °C, followed by dropwise addition of 4 N HCl (1.5 mL). The solution was warmed to 23 °C and stirred at this temperature for 4 h. The solution was cooled to 0 °C, and ethyl acetate was added. Solid NaHCO_3 was added to quench the reaction until no bubbles evolved. Then anhydrous Na_2SO_4 was added, and the solid was filtered and washed with ethyl acetate. The solvent was removed under vacuum, and the crude product was purified by flash chromatography on silica gel ($\text{CHCl}_3/\text{CH}_3\text{OH}$ 95/5) to afford the tetraol (2.1 mg).

The crude product (2.1 mg) was dissolved in ethyl acetate (1 mL) and methanol (1 mL) followed by addition of 10% Pd/C (7 mg) and formic acid (50 μL). The mixture was stirred at 23 °C for 1 h. Then Pd/C was filtered with Celite, washed with ethyl acetate, and concentrated under vacuum. The crude product was purified by flash column chromatography on silica gel ($\text{CHCl}_3/\text{MeOH}$ 93/7 to 9/1) to afford peloruside B (**2**) (1.3 mg, 24% yield over three steps). The synthetic peloruside B was identical to the natural peloruside B by comparison of 1D and 2D NMR data: $[\alpha]_{\text{D}}^{25}$ $= 0$ (c 0.09, CH_2Cl_2); ^1H NMR (600 MHz, CDCl_3) δ 0.86 (t, $J = 7.5$ Hz, 3H, H_3 -20), 1.09 (s, 3H, H_3 -21), 1.11 (s, 3H, H_3 -22), 1.17 (m, 1H, H-19a), 1.40–1.46 (m, 2H, H-12a, H-19b), 1.53 (m, 1H, H-6a), 1.72 (s, 3H, H_3 -23), 1.77 (ddd, $J = 2.4, 5.1, 12.5$ Hz, 1H, H-6b), 1.84 (m, 1H, H-4a), 1.98–2.22 (m, 3H, H-4b, H-12b, H-14a), 2.56 (m, 1H, H-18), 2.93 (br s, 1H, OH), 3.13 (br s, 1H, OH), 3.35 (m, 1H, H-24a), 3.38 (s, 3H, 7-OMe), 3.48 (s, 3H, 13-OMe), 3.63 (dd, $J = 3.6, 10.8$ Hz, 1H, H-24b), 3.79 (ddd, $J = 3.0, 5.0, 11.5$ Hz, 1H, H-7), 3.97–4.01 (m, 2H, H-8, H-13), 4.15 (m, 1H, H-5), 4.41 (br s, 1H, OH), 4.50 (br s, 1H, H-2), 4.58 (br s, 1H, H-3), 4.91 (br d, $J = 10.6$ Hz, 1H, H-11), 5.02 (d, $J = 10.3$ Hz, 1H, H-17), 5.70 (d, $J = 10.6$ Hz, 1H, H-15), 6.76 (br s, 1H, 2-OH); ^{13}C NMR (150 MHz,

CDCl_3 δ 12.4 (C-20), 15.8 (C-21), 17.8 (C-23), 21.0 (C-22), 24.7 (C-19), 31.6 (C-6), 33.9 (C-12), 35.7 (C-14), 38.6 (C-4), 43.6 (C-18), 43.8 (C-10), 55.8 (C-7-OMe), 59.3 (C-13-OMe), 63.8 (C-5), 67.0 (C-8), 67.1 (C-24), 69.91 (C-2), 69.94 (C-3), 71.8 (C-15), 74.0 (C-11), 76.1 (C-7), 78.1 (C-13), 102.1 (C-9), 131.0 (C-17), 136.9 (C-16), 174.7 (C-1); see Supporting Information for 2D NMR spectra; FT-IR (film, NaCl) 1084, 1462, 1736, 2850, 2919 cm^{-1} .

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Supporting Information Available: Full experimental details for compounds **5** and **9–12** and characterization of selected compounds; copies of ^1H and ^{13}C NMR spectra of selected compounds; comparison of ^1H and ^{13}C NMR spectra of synthetic and natural peloruside B. This material is available free of charge via the Internet at <http://pubs.acs.org>.